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Class Test 1

60 min.

Special Relativity & Intro to Quantum Mechanics

1. Two electrons in a collider experiment are approaching each other from opposite directions. An observer in the lab measures their velocities to be $0.750c$ and $-0.850c$ respectively. What is the velocity of the one electron as seen from the other ? (8)

2. (a) Suppose that a particle of mass m initially at rest has a force acting on it while it moves a distance s and derive the expression

$$E_k = \int_0^v v d(\gamma m_0 v)$$

for the work done on the particle to bring it to a final velocity v . (5)

- (b) Use integration by parts ($\int x dy = xy - \int y dx$) to derive the relativistic expression for the kinetic energy

$$E_k = mc^2 - m_0c^2.$$

Hint : $\int \frac{x dx}{\sqrt{1-x^2}} = -(1-x^2)^{1/2}$ (5)

- (c) Name the two terms on the right hand side and discuss their significance. (4)
- (d) Evaluate the second term for an electron of mass 9.11×10^{-31} kg in units of electron volts. (5)
- (e) Suppose an electron were accelerated by a potential difference of 1.0 kV. Evaluate the Lorentz factor, γ , for the electron working from the expression for the kinetic energy of the electron derived above. Comment on whether such an electron would be regarded as relativistic or not. (5)
- (f) Show that the normal expression for the kinetic energy is recovered for low velocities. Hint : Binomial Expansion : $(1+x)^n \approx 1+nx$ (8)

[32]

3. Consider the “particle in a box” problem from a quantum mechanical perspective. The un-normalised wave function for a particle in a box is

$$\psi_n(x) = A \sin \frac{\sqrt{2mE_n}}{\hbar} x = A \sin \frac{n\pi x}{L} \quad n=1,2,3,\dots$$

- (a) The normalisation condition is chosen so that $|\psi(x)|^2$ represents a probability density. Perform this normalisation and show that the normalised wave function is

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin \frac{\sqrt{2mE_n}}{\hbar} x = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L} \quad n=1,2,3,\dots$$

Hint : $\sin^2 \theta = \frac{1}{2}(1 - \cos 2\theta)$ (6)

- (b) Show, with a clearly argued derivation, that quantisation of the energy of the particle in a box arises naturally by application of the boundary conditions leading to the energy spectrum

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \quad n=1,2,3,\dots \quad (6)$$

- (c) Can the particle in the box ever have no kinetic energy ? Explain why. (2)
- (d) Sketch the wave functions and the corresponding probability densities for the first three energy levels for the particle in the box. Label the diagram clearly. (6)

[20]

Total Marks [60]