



Wits Campus, Empire Road, Johannesburg  
 PO WITS  
 South Africa  
 2050

Telephone : (011) 711-6923  
 Direct : (011) 717-6928  
 Fax : (011) 717-6932  
 e-mail : connell@inkosi.src.wits.ac.za

Quantum Mechanics of Atoms, Statistical Mechanics, Solid State Physics, Semiconductivity

## Class Test 2

**60 min.**

1. (a) From the table of the normalised wave functions for the hydrogen atom, write down the wave function for the  $2p$  electron with no  $z$ -component of orbital angular momentum. (2)

- (b) Taking note that the angular parts of the wavefunction are separately normalised, show that the expression for the electron's radial probability distribution is

$$P(r)dr = \frac{r^4}{8a_0^5} e^{-r/a_0}.$$

Note that the volume element in spherical co-ordinates is  $dV = r^2 \sin \theta dr d\theta d\phi$  and that  $\psi(r, \theta, \phi) = R(r)\Theta(\theta)\Phi(\phi)$ .

(4)

- (c) Sketch this radial probability distribution.

(4)

- (d) Show that the most probable value of  $r$  for this electron is  $4a_0$ .

(5)

[15]

2. The density of aluminium (group III of the Periodic Table) is  $2.70 \text{ g/cm}^3$  and its atomic mass is  $26.97 u$ .

- (a) Calculate the Fermi energy in aluminium. (6)

- (b) The measured Fermi energy is  $11.8 \text{ eV}$ . Find the effective mass of a conduction electron in aluminium. (6)

- (c) Explain the physical origins of effective mass. (6)

- (d) Show that electronic density of states

$$n(\epsilon)d\epsilon = \frac{8\sqrt{2}\pi V m^{3/2}}{h^3} \frac{\sqrt{\epsilon}d\epsilon}{e^{(\epsilon-\epsilon_F)/kT} + 1}$$

may be expressed as

$$n(\epsilon)d\epsilon = (3N/2)\epsilon_F^{-3/2}\sqrt{\epsilon}d\epsilon$$

at 0 K.

(6)

- (e) Now show that the total energy of the electron gas at 0 K is

$$E_0 = \frac{3}{5}N\epsilon_F$$

by evaluating

$$E_0 = \int_0^{\epsilon_F} \epsilon n(\epsilon) d\epsilon.$$

and hence arrive at the result

$$\bar{\epsilon}_0 = \frac{3}{5} N \epsilon_F \tag{6}$$

- (f) If electrons behaved classically, what would the temperature of the metal have to be for a similar average energy to of the electrons to be realised. (6)

[24]

Total Marks [50]